Experiment

Synthesis of Tolbutamide

Aim;- To synthesize Tolbutamide and calculate its percentage yield.

Requirement;-

- Chemicals- p-Toluene Sulfonamide 1 g, Butyl isocyanate 1 g, Triethylamine 1.2 ml Tetrahydrofuran 10 ml, Diethyl ether.
- Glassware- round bottom, Measuring cylinder, glass rod, ice bath

Principle;-

The synthesis of tolbutamide involves addition reaction of p-toluene sulfonamide and butyl isocyanate in the presence of triethylamine and tetrahydrofuran.

$$CH_3$$
— SO_2 - NH_2 + C_4H_9 - N = C = O
 CH_3
 CH_3

Procedure;-

- 1. About N-butyl isocyanate (1m mol) and triethylamine (1.2m mol) in a round bottom flask containing 10 ml of tetrahydrofuran, kept in an ice bath.
- 2. To the above mixture P-toluene sulfonamide (1m mol) was added drop wise at 0°c.
- 3. After completing the addition, the temperature was suddenly raised to 35-45oC and stirred for 3-4 Hrs.
- 4. Then the solution was filtered. The product was separated and dried. Then it was recrystallized by using Diethyl ether.

Calculation:-

Theoretical yield:-....g

Practical yield:-....g

Formula:- % yield = (Practical yield)/(Theoretical yield)×100

Result:- The Tolbutamide was synthesized & percent yield was found to be%.